OUTLINE

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Inter-band & Sub-band Quantum Cascade Laser based Trace Gas Sensor Technologies: Recent Advances and Applications

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• New Laser Based Trace Gas Sensor Technology

- Novel Multipass Gas Absorption Cells & Electronics
- Quartz Enhanced Photoacoustic Spectroscopy

• Examples of nine mid-infrared Trace Gas Species

C₂H₆, NH₃, NO, CO, SO₂, CH₄, N₂O, H₂O₂ & C₃H₆O
 Future Directions of Laser Based Gas Sensor Technology and Conclusions

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Wide Range of Trace Gas Sensing Applications

- Urban and Industrial Emission Measurements
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions

Rural Emission Measurements

• Agriculture & Forestry, Livestock

Environmental Monitoring

- Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
- Volcanic Emissions

Chemical Analysis and Industrial Process Control

- Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- Spacecraft and Planetary Surface Monitoring
 - Crew Health Maintenance & Life Support
- Applications in Medical Diagnostics and the Life Sciences
- Technologies for Law Enforcement, Defense and Security
- Fundamental Science and Photochemistry



"Curiosity" Landed on Mars on August 6, 2012



Laser based Trace Gas Sensing Techniques

- Optimum Molecular Absorbing Transition
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- Long Optical Pathlength
 - Multipass Absorption Cell (White, Herriot, Chernin, Sentinel Photonics/Aeris Technologies)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- Spectroscopic Detection Schemes
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)



- Faraday Rotation Spectroscopy (limited to paramagnetic chemical species)
- Differential Optical Dispersion Spectroscopy (DODiS)
- Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)
- Frequency Comb Spectroscopy
- Laser Induced Breakdown Spectroscopy (LIBS)



HITRAN Simulated Mid-Infrared Molecular Absorption Spectra



Source: HITRAN 2012 database

Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to pptv)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Mode Hop-free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	High wall plug efficiency, no cryogenics or cooling water
Field deployable in harsh environments	Compact & Robust

Key Characteristics of Mid-IR QCL & ICL Sources – Sept. 2014

• <u>Band – structure engineered devices</u>

Emission wavelength is determined by layer thickness – MBE or MOCVD; Type I QCLs operate in the 3 to 24 μ m spectral region; Type II and GaSb based ICLs can cover the 3 to 6 μ m spectral range.

- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

• Wide spectral tuning ranges in the mid-IR

- 1.5 cm⁻¹ using injection current control for DFB devices
- 10-20 cm⁻¹ using temperature control for DFB devices
- ~100 cm-1 using current and temperature control for QCL DFB Array
- ~ 525 cm⁻¹ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array

<u>Narrow spectral linewidths</u>

- CW: 0.1 3 MHz & <10kHz with frequency stabilization (0.0004 cm⁻¹)
- Pulsed: ~ 300 MHz

<u>High pulsed and CW powers of QCLs at TEC/RT</u> <u>temperatures</u>

- Room temperature pulsed power of > 30 W with 44% wall plug efficiency
- CW powers of \sim 5 W with 23% wall plug efficiency at 293 °K
- > 600 mW CW DFB @ 285 °K; wall plug efficiency 23% at 4.6 μ m





45 nm

Motivation for Mid-infrared C₂H₆ Detection

Application in medical breath analysis

- Asthma
- Schizophrenia
- Lung cancer
- Vitamin E deficiency

Atmospheric chemistry and climate

- Fossil fuel and biofuel consumption
- Biomass burning
- Vegetation/soil
- Natural gas loss



HITRAN absorption spectra of C_2H_6 , CH_4 , and H_2O



C₂H₆ Detection with a 3.36 μm CW DFB LD using a Novel Compact Multipass Absorption Cell and Control Electronics



Schematic of a C_2H_6 gas sensor using a Nanoplus 3.36 μ m DFB laser diode. M – mirror, CL – collimating lens, DM – dichroic mirror, MC – multipass cell, L – lens, SCB – sensor control board.



Minimum detectable C_2H_6 concentration: ~ 740 pptv (1 σ ; 1 s time resolution)



Innovative long path, small volume multipass gas cell: **57.6 m with 459 passes**



MGC dimensions: **17 x 6.5 x 5.5 (cm)** Distance between the MGC mirrors: 12.5 cm

Typical Oil & Gas Production Site near Houston, TX



This figure shows the result of a sequence of four fracking injections obtained by directional drilling which creates horizontal production in target stratum.

Proposed ARPA-E CH4 detection project at $3.327 \ \mu m$ at well platform of 10m x10m with a 1 m spatial resolution starting in 2015





Motivation for NH₃ Detection

Medical diagnostics

- Kidney disease
- Liver failure and Cirrhosis
- Brain Cells dysfunction
- Drowsiness and Coma
- Atmospheric chemistry
- Pollutant gases monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques associated with electric power plants
- Spacecraft related trace gas monitoring



Conventional Photoacoustic Spectroscopy (PAS)





NH₃ Measurements based on an EC-QCL PAS Sensor System





NH₃ sensor deployed at the UH Moody Tower rooftop monitoring site.

Schematic of a Daylight Solutions 10.36 µm CW TEC EC-QCL based PAS NH₃ Sensor.



Diurnal profile of atmospheric NH₃ levels in Houston, TX.



Comparison between NH₃ and particle number concentration time series from July 19 to July 31 2012.

Unexpected Remote Detection of NH₃ based on PAS





A chemical incident occurred at \sim 6 a.m. after two trucks collided on I-59. Both trucks caught fire. [www.chron.com]



Estimated hourly NH_3 emission from the Houston Ship Channel area is about 0.25 ton. Mellqvist et al., (2007) Final Report, HARC Project H-53.



Remote Detection of Sporadic NH₃ Emissions by the Parish Electric Power Plant, TX





The Parish electric power plant is located near the Brazos River in Fort Bend County, Texas (~27 miles SW from downtown Houston)



Fort-Worth, Dallas(TX) CAMS 75 & TCEQ Monitoring Site





Available Instrumentation at CAMS 75 & TCEQ monitoring site

Species/parameter	Measurement technique				
NH ₃	Daylight Solutions External Cavity Quantum Cascade Laser (Photo-acoustic Spectroscopy)				
СО	Thermo Electron Corp. 48C Trace Level CO Analyzer (Gas Filter Correlation)				
SO ₂	Thermo Electron Corp. 43C Trace Level SO ₂ Analyzer (Pulsed Fluorescence)				
NO _x	Thermo Electron Corp. 42C Trace Level NO-NO ₂ -NO _X Analyzer (Chemiluminescence)				
NO _y	Thermo Electron Corp. 42C-Y NO _Y Analyzer (Molybdenum Converter)				
HNO ₃	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)				
HCl	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)				
VOC _s	IONICON Analytik Proton Transfer Reaction Mass Spectrometer and TCEQ Automated Gas Chromatograph				
PBL height	Vaisala Ceilometer CL31 with updated firmware to work with Vaisala Boundary Layer View software				
Temperature	Campbell Scientific HMP45C Platinum Resistance Thermometer				
Wind speed	Campbell Scientific 05103 R. M. Young Wind Monitor	HAN .			
Wind direction	Campbell Scientific 05103 R. M. Young Wind Monitor	DE			

NH₃ Source Attribution & Temperature Variations

- Emission events from specified point sources (i.e., industrial facilities)
- Estimated NH₃ emissions from cows (1.3 tons/day)
- Estimated NH₃ emissions from soil and vegetation (0.15 tons/day)
- EPA PMF (biogenic:74.1%; light duty vehicles:12.1%; natural gas/industry: 9.4%; heavy duty vehicles:4.4%)
- Livestock might account for approximately 66.4% of total NH₃ emissions
- Increased contribution from industry (→18.9%)



30 May 2011 - 30 June 2011

From Conventional PAS to Quartz Enhanced PAS (QEPAS)



Use of Canines in non-invasive & sensitive Cancer Detection





• Advantages

- Non-invasive, safe and easy sample collecting
- Testing can be conducted several weeks / months after sampling
- Relatively easy training and interpretation of dogs' indications
- Odor samples can be tested several times
- Extremely high detection sensitivity and specificity
- Potential of VOCs in search, rescue and emergency application

Disadvantages

- To-date a "Black-box technology"
- It is a method aimed at earning a reward, which starts to occur after 4 years +
- Variation of sensitivity and specificity
- Re-training of dogs tends to be less effective



Quartz Tuning Fork as a Resonant Microphone for QEPAS





Unique Properties

- Extremely low internal losses:
 - Q~10 000 at 1 atm
 - Q~100 000 in vacuum
- Acoustic quadrupole geometry
 - Low sensitivity to external sound
- Large dynamic range (~10⁶) linear from thermal noise to breakdown deformation
 - 300K noise: $x \sim 10^{-11}$ cm
 - Breakdown: $x \sim 10^{-2}$ cm
- Wide temperature range: from 1.6K to ~700K

Acoustic Micro-resonator (µR) Tubes

- Optimum inner diameter: 0.6 mm; μR-QTF gap is <u>25-50 μm</u>
- Optimum mR tubes must be ~ $\frac{4.4 \text{ mm long}}{(\sim \lambda/4 < l < \lambda/2 \text{ for sound at } 32.8 \text{ kHz})}$
- SNR of QTF with µR tubes: <u>30</u> (depending on gas composition and pressure)

Optimum NH₃ Line Selection for a 10.34 μ m CW TEC DFB QCL



Simulated HITRAN high resolution spectra (a) 130 Torr indicating two NH_3 absorption lines of interest No overlap between NH_3 and CO_2 absorption lines was observed for the selected **967.35 cm⁻¹** NH_3 absorption line in the v₂ R-band.



QEPAS based NH₃ Gas Sensor Architecture



Real-time Exhaled Human NH₃ Breath Measurements





Airway pressure (black), CO_2 (red), and NH_3 (blue) profiles of a single breath exhalation lasting 40sec.

NH₃ concnetration [ppb]

Successful testing of a 2nd generation breath ammonia monitor installed in a clinical environment.(Johns Hopkins, Baltimore, MD and St. Luke's Hospital, Bethlehem, PA)

Minimum detectable concentration of NH_3 is: ~ 6 ppbv at 967.35 cm⁻¹ (1 σ ; 1 s time resolution) Motivation for Nitric Oxide Detection

• NO in medicine and biology

- Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
- Treatment of asthma, chronic obstructive pulmonary disease (COPD) & lung rejection
- Environmental pollutant gas monitoring
 - Ozone depletion
 - Precursor of smog and acid rain
 - NO_X monitoring from automobile exhaust and power plant emissions
- Atmospheric Chemistry



Molecular Absorption Spectra within two Mid-IR Atmospheric Windows and NO absorption @ 5.26µm



Source: HITRAN 2012 database



Emission Spectra of a 1900cm⁻¹ TEC DFB QCL and HITRAN simulated spectra of NO, H₂O & CO₂



Output power: 117 mW @ 25 C Thorlabs/Maxion



Performance of a $5.26 \ \mu m \ CW \ HHL \ TEC \ DFB-QCL$



Single frequency QCL radiation recorded with FTIR for different laser current values at a QCL temperature of 20.5°C.

CW DFB-QCL optical power and current tuning at three different temperatures.



CW TEC DFB QCL based QEPAS NO Gas Sensor



Schematic of a DFB-QCL based Gas Sensor. PcL – plano-convex lens, Ph – pinhole, QTF – quartz tuning fork, mR – microresonator, RC- reference cell, P-elec D – pyro electric detector



CW HHL TEC DFB-QCL package and IR camera image of the laser beam at 630 mA and 20.5 deg C through tubes after ADM



Compact Prototype NO Sensor (September 2012)



Performance of CW DFB-QCL based WMS QEPAS NO Sensor Platform



2f QEPAS signal (navy) and reference 3f signal (red) when DFB-QCL was tuned across 1900.08 cm⁻¹ NO line.

2f QEPAS signal amplitude for 95 ppb NO when DFB-QCL was locked to the **1900.08 cm⁻¹** line.

Minimum detectable NO concentration is: $\sim 3 \text{ ppbv} (1\sigma; 1 \text{ s time resolution})$



QCL based TDLAS Sensor for Detection of NO Emission from Cancer Cells



Schematic drawing of the sensor setup

Dependence of the TDLAS sensor signal from biological samples on the gas flow (black squares). The inset shows spectra corresponding to the data points.

Motivation for Carbon Monoxide Detection

• CO in Medical Diagnostics

Hypertension and abnormality in heme metabolism

• Public Health

 Extremely dangerous to human life even at a low concentrations. CO must be monitored at low concentration levels (<35 ppm).

• Atmospheric Chemistry

- Incomplete combustion of natural gas, fossil fuel and other carbon containing fuels.
- Impact on atmospheric chemistry through its reaction with hydroxyl (OH) for troposphere ozone formation and changing the concentration levels of greenhouse gases (e.g. CH₄).



Performance of a $4.61 \mu m$ high power CW TEC DFB QCL



CW DFB-QCL based CO QEPAS Sensor Results



2f QEPAS signal for dry (red) and moisturized (blue) 5 ppm CO:N₂ mixture near 2169.2 cm⁻¹.



Dilution of a 5 ppm CO reference gas mixture when the CW DFB-QCL is locked to the **2169.2 cm⁻¹** R6 CO line.



Minimum detectable CO concentration is:

~ 2 ppbv (1 σ ; 1 s time resolution)



CW DFB-QCL based SO₂ QEPAS Results

Motivation for Sulfur Dioxide Detection

- <u>SO₂ exposure affects lungs and causes breathing</u> <u>difficulties, bronchitis, cardiovascular disease</u>
- Currently, reported annual average atmospheric SO_2 concentrations range from ~ 1 6 ppb
- Prominent air pollutant
- Emitted from coal fired power plants (~73%) and other industrial facilities (~20%)
- In the atmosphere SO₂ converts to sulfuric acid
 → primary contributors to acid rain
- SO₂ reacts to form sulfate aerosols
- Primary SO₂ exposure for 1 hour is 75 ppb



Molecular Absorption Spectra within two Mid-IR Atmospheric Windows







2f WMS QEPAS signals for different SO₂ concentrations when laser was tuned across 1380.9 cm^{-1} line.



QEPAS based CH₄ and N₂O Gas Sensor

Motivation for CH₄ and N₂O Detection

- Medical Diagnostics
 - Nausea, blurred vision, vomiting
- Prominent greenhouse gases
- Sources: wetlands, leakage from natural gas systems, fossil fuel production and agriculture





The Analyst Aug. 2013

CH₄ Measurements performed with a DFB-QCL based QEPAS Sensor installed in the Aerodyne Mobile Laboratory (Sept 7, 2013)



Atascocita Landfill, Humble, TX 77396 CH4 Perimeter Measurements



A: 29.9599° North, 95.2334° West B: 29.9364° North, 95.2508° West C: 29.9547° North, 95.2462° West (Landfill)





Jahjah et. al., Opt. Let., 39, 957-960, 2014

Motivation of H₂O₂ Detection

- Oxidative capacity of atmosphere and balance of HO_x ;
- Acid rain formation, In-cloud oxidation of S(IV) to S(VI);
- Active agent in decontamination and sterilization systems;
- H_2O_2 in breath is a biomarker of oxidative stress;
- H_2O_2 concentration levels in Houston have not been reported despite of atmospheric conditions, such as high humidity, high solar radiation levels, and the presence of the petrochemical industry.





QEPAS based Hydrogen Peroxide (H₂O₂) Sensor System



Schematic of QCL based QEPAS sensor: ADM – acoustic detection module; CEU – control electronics unit; PC – personal computer.



H₂O₂ Exposure limit is set at 1 ppmv by OSHA Simulated spectra (HITRAN) of H_2O_2 at 296 K and 130 Torr, along with atmospheric interfering molecules of CH_4 and N_2O ; two target wavelengths at 1294.1 and 1294.9 cm⁻¹ are shown.

QEPAS Performance for Trace Gas Species (September 2014)

	Molecule (Host)	Frequency,	Pressure,	NNEA,	Power,	NEC (τ=1s),
_		cm-1	Torr	cm ⁻¹ W/Hz ^{1/2}	mW	ppmv
	O ₃ (air)	35087.70	700	3.0 10-8	0.8	1.27
	O ₂ (N ₂)	13099.30	158	4.74 10 ⁻⁷	1228	13
۲	C2H2 (N2)*	6523.88	720	4.1 10-9	57	0.03
	NH3 (N2)*	6528.76	575	3.1 10-9	60	0.06
	C2H4 (N2)*	6177.07	715	5.4 10-9	15	1.7
	CH ₄ (N ₂ +1.2% H ₂ O)*	6057.09	760	3.7 10-9	16	0.24
NIR 🖌	N2H4	6470.00	700	4.1 10-9	16	1
	H2S (N2)*	6357.63	780	5.6 10-9	45	5
	HCl (N2 dry)	5739.26	760	5.2 10-8	15	0.7
	CO2 (N2+1.5% H2O) *	4991.26	50	1.4 10-8	4.4	18
5	CH2O (N2:75% RH)*	2804.90	75	8.7 10 ⁻⁹	7.2	0.12
	CO (N ₂ +2.2% H ₂ O)	2176.28	100	1.4 10-7	71	0.002
	CO (propylene)	2196.66	50	7.4 10-8	6.5	0.14
	N ₂ O (air+5%SF ₆)	2195.63	50	1.5 10-8	19	0.007
	C2H5OH (N2)**	1934.2	770	2.2 10-7	10	90
d-IR ٦	NO (N ₂ +H ₂ O)	1900.07	250	7.5 10-9	100	0.003
	C ₂ HF ₅ (N ₂)***	1208.62	770	7.8 10-9	6.6	0.009
	NH3 (N2)*	1046.39	110	1.6 10-8	20	0.006
	SF ₆	948.62	75	2.7x10 ⁻¹⁰	18	5x10 ⁻⁵ (50 ppt)

* - Improved microresonator

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Μ

** - Improved microresonator and double optical pass through ADM

*** - With amplitude modulation and metal microresonator

NNEA - normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and τ =1s time constant, 18 dB/oct filter slope.

For comparison: conventional PAS 2.2 10^{-9} cm⁻¹W/ \sqrt{Hz} for NH₃

Merits of QEPAS based Trace Gas Detection

- Very small sensing module and sample volume (a few mm^3 to $\sim 2cm^2$)
- Extremely low dissipative losses
- Optical detector is not required
- Wide dynamic range
- Frequency and spatial selectivity of acoustic signals
- Rugged transducer quartz monocrystal; can operate in a wide range of pressures and temperatures
- Immune to environmental acoustic noise, sensitivity is limited by the fundamental thermal TF noise: k_BT energy in the TF symmetric mode
- Absence of low-frequency noise: SNR scales as \sqrt{t} , up to t=3 hours as experimentally verified

QEPAS: some challenges

- Cost of Spectrophone assembly
- Sensitivity scales with laser power
- Effect of H_2O
- Responsivity depends on the speed of sound and molecular energy transfer processes



• Cross sensitivity issues

Future Directions and Outlook

- New target analytes: formaldehyde (CH₂O), ethylene (C₂H₄), ozone (O₃) and nitrate (NO₃
- Ultra-compact, low cost, robust sensors (e.g. CH₄, NO, CO...)
- QCL based ultra-portable atmospheric carbon isotope monitor for ¹²CH₄ & ¹³CH₄
- Monitoring of broadband absorbers: acetone (C_3H_6O) : MDL of 1.5 ppm with a 7mW ICL & AM, or 20ppb with a 100mW QCL @ 8.23µm; benzene (C_6H_6) ...
- Optical power build-up cavity designs (I-QEPAS)
- THz QEPAS based sensors
- Development of trace gas sensor networks



Potential Integration of a CW DFB- QCL and QEPAS Absorption Detection Module





A. Lyakh, et al "1.6 W high wall plug efficiency, continuous-wave room temperature quantum cascade laser emitting at 4.6 μ m", Appl. Phys. Lett. **92**, 111110 (2008)

Proposed Intracavity-QEPAS (I-QEPAS) Sensor System



Optical power build up cavity can provide:

- •RT CW DFB QCL, λ=4.33 microns
- •Low noise current driver → narrow QC laser linewidth ~1 MHz
- •Bow-tie cavity →4 high reflectivity mirrors, R=99.9%
- •Electronic Control Loop + PZT driver lock of cavity resonant frequency to QCL frequency

P. Patimisco, G. Scamarcio, F.K. Tittel & V. Spagnolo, "Quartz-enhanced photoacoustic spectroscopy: a review", Sensors, 14, 6165-6206 (2014)

Comparison of I-QEPAS with Other Trace Gas Sensing Techniques



P. Patimisco, G. Scamarcio, F.K. Tittel & V. Spagnolo, "Quartz-enhanced photoacoustic spectroscopy: a review", Sensors, 14, 6165-6206 (2014)

Summary and Conclusions

- Development of robust, compact, sensitive, selective mid-infrared trace gas sensor technology based on room temperature, continuous wave DFB laser diodes and high performance QCLs for environmental monitoring and medical diagnostics.
- Interband cascade and quantum cascade lasers were used in <u>TDLAS, PAS</u> and <u>QEPAS</u> <u>based sensor platforms</u>
- Eight target trace gas species were detected with a 1 sec sampling time:
 - C_2H_6 : ~3.36 µm, detection sensitivity of 740 pptv using TDLAS
 - NH₃: ~10.4 μ m, detection sensitivity of ~1 ppbv (200 sec averaging time)
 - NO: \sim 5.26 µm, detection limit of 3 ppbv
 - CO: ~4.61 µm, minimum detection limit of 2 ppbv
 - SO₂: ~7.24 μ m, detection limit of 100 ppbv
 - CH₄ and N₂O: \sim 7.28 µm, detection limits of 13 and 6 ppbv, respectively
 - H_2O_2 : ~7.73 µm, detection limit of 75 ppb
- New target analytes: CH_2O and C_2H_6O

